

Radioisotopes of significance to environmental radioactivity

^{110m}Ag

Silver (Ag)

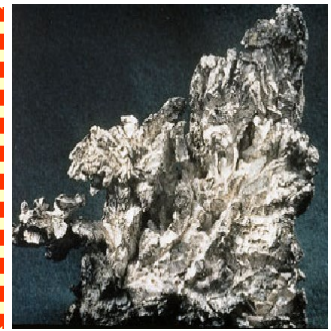
Element classification: Transition metal

No. of isotopes: 28 (2 stable, both natural)

Typical elemental concentrations:

Soil (dry): up to 44 µg/kg

Sea water: 2-100 µg/l



Behaviour in the Environment

- ◆ Metallic form and many compounds are insoluble
- ◆ Accumulates in liver and brain tissue in animals
- ◆ High transfer to fungi
- ◆ Salts are generally poorly soluble



Silver

radioecology

Key sources of radioisotopes

- ◆ Nuclear cycle: Nuclear power plants, reprocessing, waste
- ◆ Fallout: Nuclear weapons testing
- ◆ Nuclear accidents: e.g. Chernobyl, Fukushima
- ◆ Natural sources: None

Why is it of interest?

- ◆ High bioaccumulation into aquatic organisms
- ◆ High transfer to some tissues/organisms in the terrestrial environment

For more information ...

[IRSN ^{110m}Ag factsheet](#)

